

Regulatory Perspective on Safety Qualification of Extractables and Leachables

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Risk-based Approach in Evaluating E&L

- Safety considerations (e.g., toxicity, immunogenicity, etc.)
- Efficacy considerations (e.g., L interacting with a product → loss of activity; L may induce development of neutralizing activity via NAb formation)
- Quality considerations (e.g., impact on the manufacturing process, product stability, etc.)



Safety considerations

- Toxicity (e.g., acute, chronic, synergistic, additive, carcinogenicity, endocrine dysregulation, etc.)
- Adjuvant effects:
 - Adjuvants are substances that increase the activity of the immune system without having any specific antigenic effect
 - In contrast to vaccines where adjuvant effect is a desired effect, this may be a serious safety concern for therapeutic proteins
 - May promote development of anti-drug antibodies

✤A decrease or loss of efficacy due to development of neutralizing activity

- May be life-threatening if NAbs are developed against a non-redundant endogenous protein (e.g., erythropoietin - anemia/PRCA; thrombopoietin - thrombocytopenia)
- Altering the PK of the drug
- May promote non-specific inflammation

Safety considerations (cont.)

- Drug dose, mode and frequency of administration (e.g., SC vs. IV, life-time dosing and chronic exposure)
- Prior clinical exposure to leachables may enhance sensitivity in case of re-exposure
- Therapeutic necessity of the drug (higher levels may be tolerated if drug is considered a part of essential therapy)



Manufacturing considerations

- Place in the process stream (e.g., upstream vs. downstream; typically risks are greater as production moves closer to the finished product)
- Type of the processed/stored material (e.g., purification buffer vs. final product)
- Storage temperature (e.g., freezing vs. 2-8 C)
- Surface-to-volume ratio
- Contact time
- Type of polymeric material (e.g., PVC at risk for leaching di(2ethylhexyl)phthalate, which is linked to various toxicities)
- Formulation/choice of excipients; (e.g., liquid vs. lyophilized; pH; phosphate buffer)
- Risks often assessed on a case-by-case basis



What's done in practice...

- Extractables studies are performed using exaggerated conditions (organic solvents, accelerated T°, pH, etc.)
 - Alternatively, the drug manufacturer may rely on the E studies done by the vendor
 - Note: Drug Product vehicle may or may not be used as an extraction medium (role of excipients important)
- Analysis of extractables is done in conjunction with stability studies, which monitor changes in product quality over time
- Leachables studies are often omitted



Risk assessment and risk reduction with regard to leachables studies

• Safety considerations:

- Toxicity studies are usually acute studies that do not measure chronic exposure to potential leachables
- Potential for adjuvant effect and immunogenicity is not addressed
- Mode of administration (e.g., SC is often more immunogenic than IV)
- Product that is at the end of its dating period is rarely evaluated in clinic

Product Quality considerations:

- Stability studies are often not geared to detect leachable impurities (e.g., inorganic leachables such as tungsten, Fe, Al, etc., are never evaluated and organic leachables may be missed)
- Differences in the levels of leachables at the extremes of the manufacturing, storage and transportation conditions (no worst case scenario risk assessment)
- Inappropriate sample size that is needed to understand the true variability in the C/C system (e.g., tungsten and PFS syringes)



Leachables studies

- Monitor the clinical outcome in conjunction with measuring leachables over the entire shelf-life of the product
 - > In the presence of product
 - > Without the product (i.e., in placebo alone)
- Ideally, Drug Product material that is at the end of its shelf-life should be tested in clinic
- Perform stability studies, which monitor impurities and product physico-chemical and biological properties throughout the expiry period





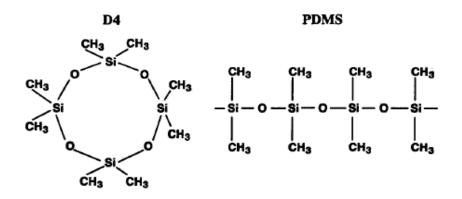
Leachables as adjuvant and/or immunomodulatory factors

Some support in the literature



Leachables as adjuvant and/or immunomodulatory factors

- Silicone oil polydimethylsiloxane and octamethylcyclotetrasiloxane (D4)
 - > Naim et al., 1995, Immunol Invest, 24:537-547
 - > Naim et al., 2000, Clin Diagnostic Lab Immunol, 7:366-370
 - Locatelli et al., 2004, Nephrol Dial Transplant, 19:288-293





Leachables as adjuvant and/or immunomodulatory factors (cont.)

 Di-(2-ethylhexyl) phthalate (DEHP) and Mono-2-ethylhexyl phthalate (MEHP)

> Larsen et al., 2001, Tox Letters, 125:11-18

> Larsen et al., 2001, Toxicology, 169:37-51

> Larsen et al., 2007, Tox Letters, 170:223-228

• Polycyclic aromatic hydrocarbons (PAH)

Lovik et al., 1997, Toxicology, 121:165-78
Nilsen et al., 1997, Toxicology, 124:225-232



Leachables as adjuvant and/or immunomodulatory factors (cont.)

- Soluble iron
 - > Beck-Speier et al., 2009, Particle and Fibre Toxicology, 6: 34-46
- Cadmium
 - Marth et al., 2001, Inter J Occupational Med and Environ Health, 14:375-386
- Nickel
 - Schmidt et al., 2010, Nature Immunology, 11:814-820
- Alkyl phenols
 - > Yano et al., 2003, J Health Sci, 49:195-204





Establishing threshold levels for leachables in biologics:

Known's and Unknown's



What we know

- Toxicological threshold levels have been proposed and/or established (e.g., PQRI, ICH Q3C, published literature, etc.) that can be applied across board
- What we don't know: Can the same approach be applied to biologics?



A feasibility exercise

- Threshold for the adjuvant effect:
 - Adjuvant effects of leachables may be studied in animal models (e.g., in mice)
 - Such studies may be useful in looking at relative differences (e.g., after a change in the C/C system) and in identifying potential risks
 - However, the threshold levels identified in animal studies are unlikely to be predictive of the clinical outcome (e.g., mice are 1,000x less sensitive to LPS compared to humans)
- Threshold for product quality
 - Needs to be assessed on a case-by case basis due to diversity of protein products, formulation composition and C/C systems
 - E.g., tungsten oxides had a very different effect on two analogous API that had different formulation: in one case tungsten caused unfolding and aggregation, whereas in another case, it had no effect



Biologics may deserve a special consideration for the following additional reasons...

- Manufacturing and stability issues:
 - Protein conformation (e.g., secondary, tertiary) is sensitive to external environment
 - Aggregation and/or degradation
 - Deamidation and/or oxidation
 - Changes in glycosylation
- Routine analytical testing often doesn't detect finite changes in the protein (e.g., release testing is unlikely to detect areas of protein unfolding unless it impacts the function)
- Large size (e.g., MAb 150 KD) and extensive surface area ensures → high frequency of potential sites of interaction
- Proteins may be more efficient in solubilizing leachables due to abundance of both hydrophilic and hydrophobic sites (the latter are usually buried in the interior of the protein)
- Drug dose, mode and frequency of administration (e.g., many biologics are sterile injectables administered frequently at relatively high volumes and doses of mg/ml)



Case studies



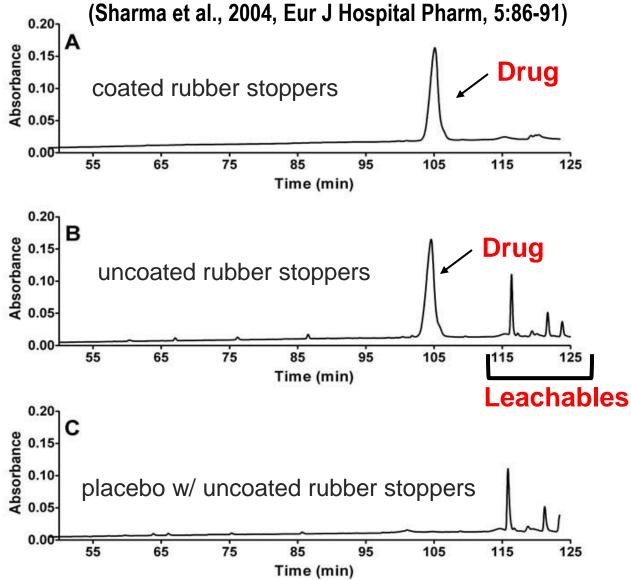
Leachables from the uncoated stoppers

(Casadevall et al., 2002, N Engl J Med, 346:469-475; Sharma et al., 2004, Eur J Hospital

Pharm, 5:86-91)

- Change from HSA to polysorbate formulation
- C/c system: pre-filled syringes with uncoated rubber stoppers
- Source: Vulcanizing agents leached from the rubber stopper during storage (e.g., Vultac 2 di, tri, tetra, penta, hexasulfide, etc.)
- Impact:
 - > no notable changes in protein physico-chemical properties
 - > safety: serious adverse event (pure red cell aplasia, PRCA)
- Hypothetical MoA: leachables acted as adjuvants leading to formation of neutralizing Abs to endogenous protein
- Resolution:
 - Switch to teflon-coated stoppers
 - > Stricter control of the cold-chain from manufacture to administration
 - S.c. route of administration was contraindicated in CRF patients, which was subsequently reversed







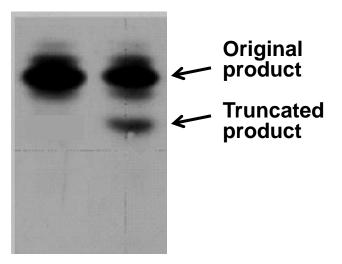
Fe leachables cause formation of proteinpreservative adducts

- Change: extension of the expiry period from 15 to 18 months
- Source: uncoated rubber stoppers released iron at levels <1 ppm
- Critical excipients: preservative and other components
- Impact:
 - Fe catalyzed oxidation of the preservative + additional excipient triggering formation of the protein-preservative adducts
 - Several sites on the protein were modified primarily at the N-terminus (primary targets were peptides with OH, NH₃ and –SH groups)
 - OOS result for protein content (e.g., >50% of the product was modified)
 - moderate decrease in potency
- Action: >10 DP lots were recalled due to OOS results
- Resolution:
 - Return to the original dating period of 15 months
 - Implement Teflon coated stoppers
 - Conduct additional studies to determine the impact on product Q and S
 - Eventually, the problematic presentation was removed from the market



Metal leachables cause product truncation via metalloprotease activation

- Change: from lyophilized to a liquid formulation
- Source: rubber stopper released divalent metal cations
- Uncovered during stability study under inverted conditions
- Mechanism: activation of a metalloprotease (process-related impurity co-eluted with API)
- Impact: product truncation at the N-terminus
- Resolution: chelator (EDTA) added to DP formulation buffer
- Adverse outcome: new formulation led to cardiovascular adverse events and a change in PK values; it was withdrawn from the market and replaced with the original one; Teflon coated stoppers implemented





Tungsten leachables from PFS barrels #1

- Container closure system: prefilled syringes
 - Tungsten filaments are used to perforate syringe barrel onto which a needle is attached
- Source: tungsten salts and tungsten oxides are deposited on the glass and into the product when contacted with liquid
- Impact: tungsten caused unfolding and aggregation of the protein
- Clinical outcome: Patients developed neutralizing Abs to the endogenous protein
- Resolution: Continue product development in vials and discontinue PFS





Tungsten leachables from PFS barrels #2

- Container closure system: prefilled syringes
- Impact: tungsten salts caused protein oxidation followed by aggregation

> Up to 60% of aggregated product found in some syringes

> Up to 1% PFS tested positive for aggregates

- Resolution (different approaches were used by different Sponsors):
 - > Optimal switch to platinum instead of tungsten filaments
 - Alternative establish tungsten specifications, nitrogen overlay process, special washing procedure, etc.



Alkali oxide extractables cause delamination of glass vials

- Container closure system: Type I borosilicate glass vials
- Glass pitting and surface delamination is initiated by ion exchange at the glasssolution interface causing breaking of Si-O bonds and weakening of the surface layer
- Risk factors promoting delamination (not listed in the order of importance):
 - > Glass vials with high surface alkalinity have higher propensity for delmination
 - Specific vial manufacturing process
 - Drug solutions formulated at alkaline pH
 - Certain buffers
 - High ionic strength
 - > Contact time (delamination is a time-dependent process) and temperature
- Risk mitigation strategies include the following:
 - > Switch to highly resistant glass (high coefficient of thermal expansion)
 - Switch to a different vial manufacturing process with more stringent temperature control (leads to lower surface alkalinity)
 - > Switch to lower risk formulation (e.g., pH, buffers, etc.)
 - Reduce product dating period



Aluminum leachables from glass form particulates

- Change: change from molded to tubing glass vials
- Source: aluminum oxide leached from the new glass vials
- Mechanism: aluminum interacted with sodium phosphate in the formulation forming aluminum phosphate crystals
 - Aluminum phosphate (i.e., alum) is widely used as adjuvant in vaccines, although at concentration leached, it is unlikely to exert such an effect
- Impact: visible particles (up to 150 µm diam) observed in stability samples over 12 month of age with no other OOS results
- Resolution:
 - Recall of lots that failed the particulate spec
 - New glass vials are coated using a baked-on siliconization process



Barium leachables from glass form particulates

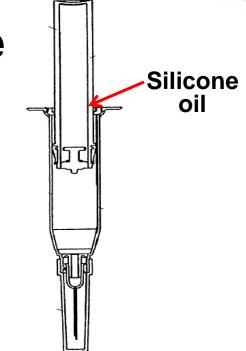
- Change: vendor for glass vials
- Source: barium leached from new glass vials
- Mechanism: barium interacted with sodium sulfate in the formulation forming barium sulfate crystals
- Impact: visible particles observed in 18 month stability samples with no other OOS results
- Resolution: acceptance limit for barium established with commitment to generate stability data for 10 new vial lots

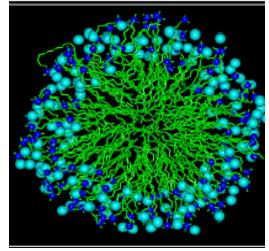
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Silicone oil leachables cause product aggregation

- Source: silicone oil spray was used as a lubricating agent to coat prefilled syringes
- Observation: silicone oil was shed (break-loose effect) from the syringe barrel into the product
- Silicone oil forms micelles in solution, which can interact with proteins and cause protein denaturation and aggregation (J Pharm Sci, 2005, 94:918-927)
- Outcome: formation of amorphous polymers visible by the naked eye
- Resolution: product now packaged in nonsiliconized syringes







Points to consider...

- Greater emphasis should be placed on the leachables testing, in conjunction with clinical data mining, in order to reduce clinical uncertainty and minimize patients' exposure to unnecessary risks
- Consideration for the improvement and standardization of E&L testing
- Consideration of E&L for disposable (i.e., single use) systems



Summary

- Biologic products can be sensitive to minor impurities and changes in the C/C system and/or formulation
- Undetected differences in product impurities may have a significant impact on clinical safety and efficacy (e.g., leachables acting as adjuvants triggering NAb response)
- It's important to monitor leachables over time (e.g., extended time points reflective of product dating period should be included)
- Corrective actions should employ a simplest approach to resolve a problem while minimizing changes to product quality as it relates to safety and efficacy
- Lack of transparency between a product quality attribute and its safety & efficacy doesn't reduce risk, it's simply, uncontrolled risk (Barry Cherney)



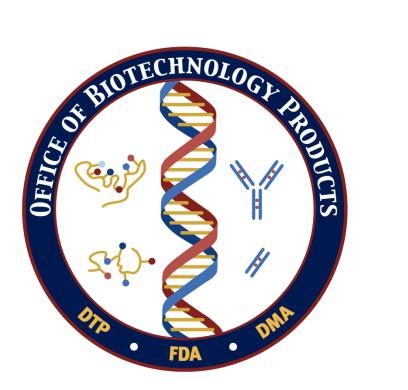
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Questions? Comments?



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Extractables analysis

- Appropriate extraction procedure is relevant
 - Determines chemical profile and maximal levels of extractables (depends on the specific extraction conditions)
 - Highlights immediate safety concerns (if any)
 - Ensures that methods with appropriate specificity and sensitivity are used
- Justification of selected extraction conditions is generally poorly described in submissions



Examples of more common E&L

- Phthalates (e.g., Di(2-ethylhexyl) phthalate DEHP)
- Metals (e.g., Zn, Fe, Ba, Ca, Al, Ni, etc.)
- Fatty acids (e.g., stearic, plamitic myristic, etc.)
- Cyclic esters (from polyurethane adhesives)
- Silicone oil (e.g., polydimethylsiloxane)
- Organic solvents (e.g., acetone, isopropanol, etc.)
- Nitrosamines (e.g., diphenylnitrosamines, etc.)
- Vulcanizing agents (e.g., Vultac 2, etc.)
- Accelerators (e.g., thiuram, sulfenamide, guanidine, dithiocarbamate, etc.)
- Antioxidants (e.g., BHT, Irganox, Irgafos, etc.)
- Polycyclic aromatic hydrocarbons
- Antistatic agents
- Cleaning agents